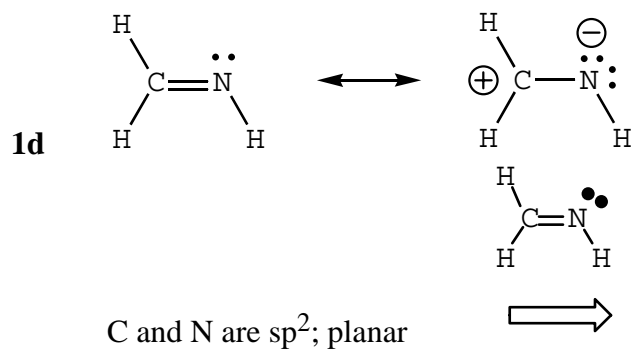
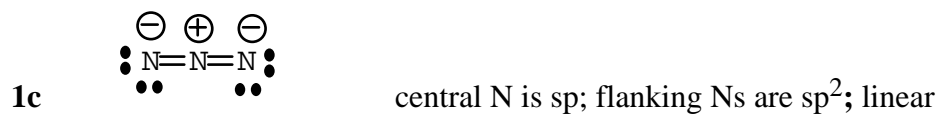
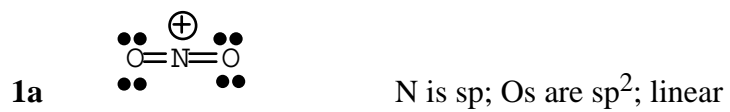
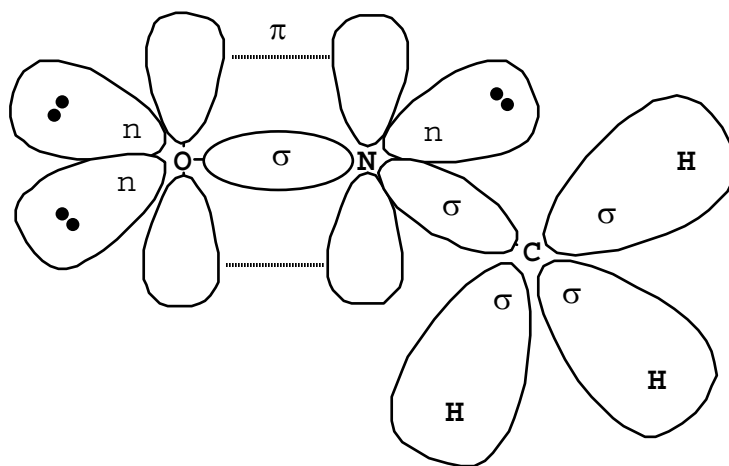


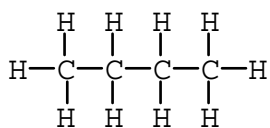
Answers



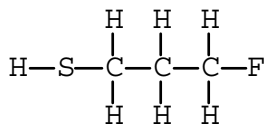
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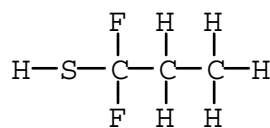
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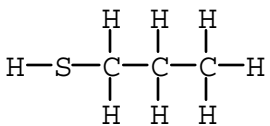
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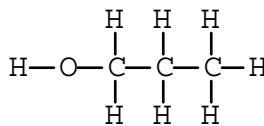
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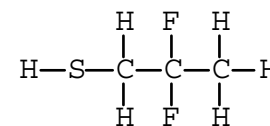
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4



5



2

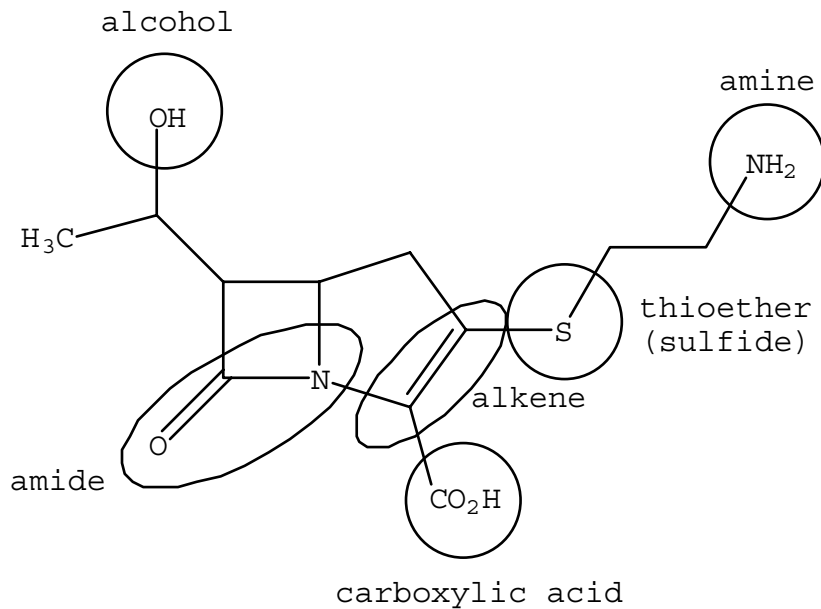
4a $\text{CH}_3\text{S}^- + \text{H}_2\text{O}$ eq. to the right

4b $(\text{CH}_3)_3\text{CC}\equiv\text{C}^- + (\text{CH}_3)_2\text{NH}$ eq. to the right

4c $(\text{CH}_3)_3\text{CC}\equiv\text{C}^- + (\text{CH}_3)_3\text{COH}$ eq. to the left

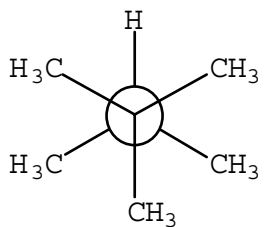
4d $\text{Cl}^- + \text{CH}_4$ eq. to the right

5

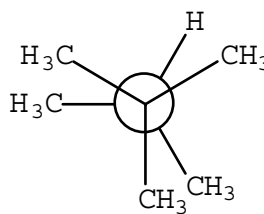


- 6a** 1,1,1-trifluoro-2-methylbutane
6b 4-ethyl-1,2-dimethyl-6-(2-methylpropyl)cycloheptane
 or 4-ethyl-6-isobutyl-1,2-dimethylcycloheptane
6c 1-bromo-2,3,5-trimethylheptane
6d *cis*-1-methyl-4-(1-methylethyl)cyclohexane
 or *cis*-1-isopropyl-4-methylcyclohexane

7

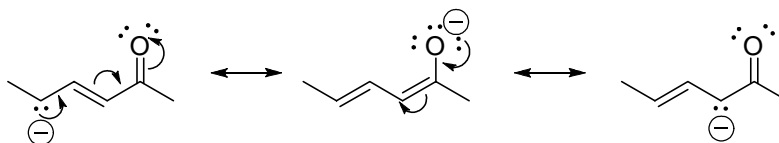
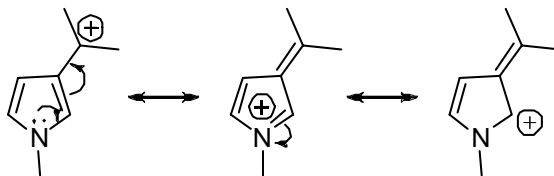
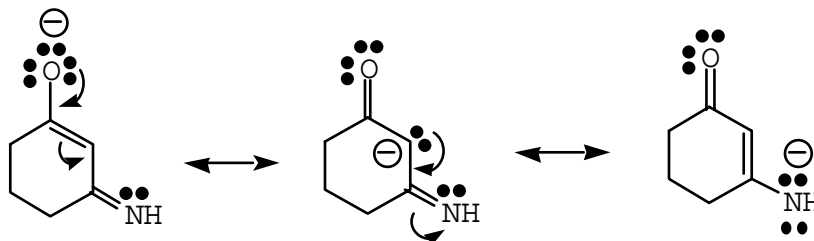


most stable
 $E_{strain} = 3.6 \text{ kcal/mol}$

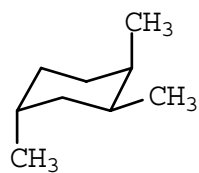


least stable
 $E_{strain} = 6.6 \text{ kcal/mol}$

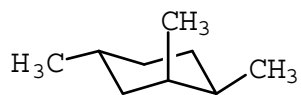
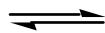
8



9



$$E_{strain} = 4.5 \text{ kcal/mol}$$



$$E_{strain} = 2.7 \text{ kcal/mol}$$