Scratch paper

Do not write any answers you wish to be graded on this page

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USEFUL INFORMATION

Partial periodic table

Atomic numbers and electronegativities

| IA | IIA | IIIA | IVA | VA | VIA | VIIA | VIIIA |
|------|------|------|------|------|------|------|-------|
| 1 | | | | | | | 2 |
| Н | | | | | | | He |
| 2.20 | | | | | | | _ |
| 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| Li | Be | В | С | N | 0 | F | Ne |
| 0.98 | 1.57 | 2.04 | 2.55 | 3.04 | 3.44 | 3.98 | |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| Na | Mg | ΑI | Si | Р | S | CI | Ar |
| 0.93 | 1.31 | 1.61 | 1.90 | 2.19 | 2.58 | 3.16 | |
| 19 | 20 | 31 | 32 | 33 | 34 | 35 | 36 |
| K | Ca | Ga | Ge | As | Se | Br | Kr |
| 0.82 | 1.00 | 1.81 | 2.01 | 2.18 | 2.55 | 2.96 | |
| 37 | 38 | 49 | 50 | 51 | 52 | 53 | 54 |
| Rb | Sr | In | Sn | Sb | Te | I | Xe |
| 0.82 | 0.95 | 1.78 | 1.96 | 2.05 | 2.1 | 2.66 | |
| 55 | 56 | 81 | 82 | 83 | 84 | 85 | 86 |
| Cs | Ва | TI | Pb | Bi | Po | At | Rn |
| 0.79 | 0.89 | 2.04 | 2.33 | 2.02 | 2.0 | 2.2 | _ |

| Type of hydrogen | δ [ppm] | Type of hydrogen | δ [ppm] | Type of hydrogen | δ [ppm] |
|---------------------|----------|--------------------------|---------|------------------|---------|
| alcohol H -0 | variable | H — C≡C | 2–3 | H -C-O | 3.2–3.8 |
| H — N | variable | H —C—N | 2.2-2.8 | H — C=C | 4–7 |
| H— R | 0.8–1.6 | H -C- | 2.2 | н— | 6.5–8.5 |
| H —C—C=C | 1.6 | H-C-X (X = C1, Br, I) | 2.2–4.2 | H —C=O | 8–10 |
| H —C—C=O | 2.0-2.2 | | | acid H -0 | 10–14 |

CAS CH 203 Final Examination

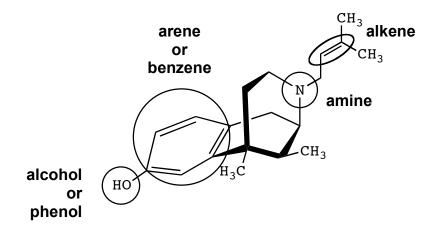
Organic Chemistry I 16 Dec 2010, 12:30 P.M.–2:30 P.M.

| Name | |
|-------------------|------------------------------------|
| | |
| ID Number | |
| | |
| Instructions | |
| | |
| (A) Make sure | you have 6 pages with 9 questions. |
| (D) Write all an | awara an the nagge provided |
| (b) write all ans | swers on the pages provided. |

- (C) Only answers written in ink will be considered for regrading.
- (D) Good luck!

| Page | Score |
|-------|-------|
| 1 | /22 |
| 2 | /19 |
| 3 | /12 |
| 4 | /27 |
| 5 | /10 |
| 6 | /12 |
| Total | /100 |

(1) Circle and name the functional groups in the obstetric anesthetic pentazocine. (4 points)



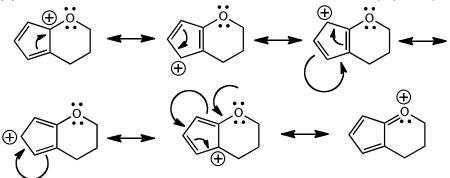
(2) Write the systematic name of these compounds; include stereochemical designations where appropriate. (9 points)

dibromo-3-methylcycloheptane

(b)
$$CH_3$$
 CH_3 (E)-2,6-dimethylocta-2,6-diene or (6E)-2,6-dimethyl-2,6-diene or cotadiene

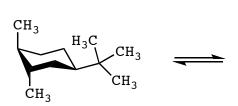
(c)
$$H_3C$$
 CH_3 (2,2-dimethylbutyl)cyclopentane or 1-(2,2-dimethylbutyl)cyclopentane

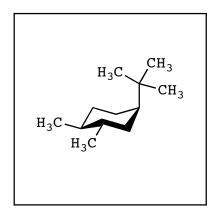
(3) Draw three additional resonance structures of this ion. (9 points)



- **(4)** The chair conformation shown below is in equilibrium with a second chair.
- (a) In the box provided draw the second chair.(4 points)
- **(b)** Using data in the table calculate the strain energy (E_{strain}) of both chairs; write your answers in the space provided. (6 points)

| Interaction | Estrain | | |
|---|------------|--|--|
| | [kcal/mol] | | |
| CH ₃ ,CH ₃ gauche | 0.9 | | |
| CH ₃ ,H 1,3-diaxial | 0.9 | | |
| C(CH ₃) ₃ ,H 1,3-diaxial | 2.7 | | |





Estrain = 3.6 kcal/mol

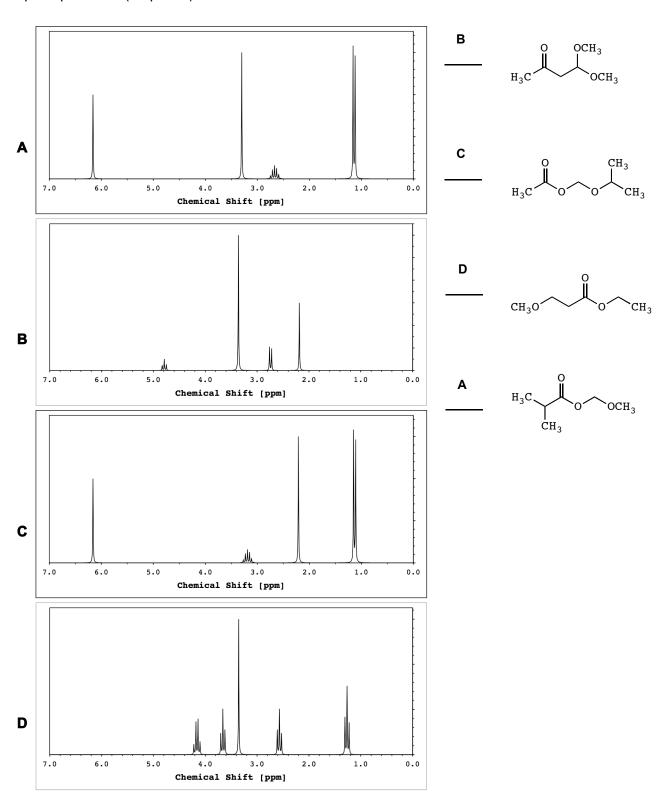
Estrain = 6.3 kcal/mol

(5) Write a mechanism of this reaction; use curved arrows to show the movement of electrons. (9 points)

$$H_{3}C$$

$$H$$

(6) Match NMR spectra **A–D** to the appropriate compound at right; write your answer in the space provided. (12 points)



(7) Draw the structure of the major organic product of these reactions and reaction sequences; in those reactions marked §, clearly show the stereochemistry of the product using dashing-and-wedging. Do not present mechanisms. *Note* Reaction sequence (g) produces two organic products: draw both of them. (27 points)

(a)
$$H_3C$$
 OH + NaOH \longrightarrow H_3C O O O

(b§)
$$\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}}{\overset{\mathbb{C}_1}}{\overset{\mathbb{C}_1}}}}}}}}}}}}}}}}}}}}}$$

(c)
$$\frac{\text{Cl}_2}{\text{light}}$$

(d§)
$$H_2$$
 H_2 H_2

(e§)
$$H_3C$$
 CH_3 $EtoH$ H_3C CH_3 H_3C CH_3 H_3C CH_3

(f)
$$H_3C$$
 Br H_3C OCH₃ CH_3 CH_3

(g)
$$H_3C$$
 $=$ $(1) \text{ NaNH}_2$ H_3C $+$ CH_3 CH_3 CH_3 CH_3

(h)
$$H_3C$$
 \longrightarrow CH_3 H_2 \longrightarrow H_3C \longrightarrow CH

(8) The reaction shown below takes place by a radical mechanism. In the spaces provided present the mechanism: write initiation, propagation, and two termination steps. (10 points)

Initiation

Propagation

Termination (write two termination steps)

$$F \cdot \bigcap_{F} F \longrightarrow F_{2}$$

$$\bigcap_{+} F \cdot \bigcap_{F} F$$

$$\bigcap_{+} F \cdot \bigcap_{F} F$$

(9) Propose a sequence of reactions that efficiently converts the indicated starting material(s) to the target molecule. Show the reagents needed in each step and draw the product formed after each step. Do not present mechanisms. (6 + 6 = 12 points)

(a)
$$CH_3$$
 CH_3

$$CH_3$$
 KCN
 CH_3
 CN

(b)
$$H_3C$$
 + Br CH_3 H_3C CH_3