

Table 1: Fundamental and derived constants

Name	Symbol	Value	Units
Atomic Mass Unit	u	$1.660\,54 \times 10^{-27}$	kg
Avogadro's Number	N_A	$6.022\,14 \times 10^{23}$	mol ⁻¹
Mass of an Electron	m_e	$9.109\,38 \times 10^{-31}$	kg
Mass of a Neutron	m_n	$1.674\,93 \times 10^{-27}$	kg
Mass of a Proton	m_p	$1.672\,62 \times 10^{-27}$	kg
Faraday's Constant	F	$9.648\,53 \times 10^4$	C/mol
Gas Constant	R	8.314 46	J/(K mol)
		0.082 06	L atm/(K mol)
		62.363 58	L Torr/(K mol)
Boltzmann's Constant	k_B	$1.380\,65 \times 10^{-23}$	J/K
Planck's Constant	h	$6.626\,07 \times 10^{-34}$	J s
Speed of Light	c	$2.997\,92 \times 10^8$	m/s
Earth's Gravitational Constant	g	9.806 65	m/s ²
Bohr Radius	a_0	$5.291\,77 \times 10^{-11}$	m
Rydberg Constant	R_∞	$1.097\,37 \times 10^7$	m ⁻¹
Hartree Energy	E_h	$4.359\,74 \times 10^{-18}$	J

Table 2: Formulas

$\ln 2 = 0.693$	$c_V = \frac{3}{2}R$	$\Delta S_{\text{sys}} = nc_P \ln \left(\frac{T_2}{T_1} \right)$
$\ln 3 = 1.099$	$\Delta U = nc_V \Delta T$	$\Delta S_{\text{sys}} = nc_V \ln \left(\frac{T_2}{T_1} \right)$
$\ln 5 = 1.609$	$\Delta H = nc_P \Delta T$	$\int_0^\infty e^{-\alpha x} dx = \frac{1}{\alpha}$
$\ln N! \approx N \ln N - N$	$w = - \int_{V_1}^{V_2} P_{\text{ext}}(V) dV$	$\int_0^\infty e^{-\alpha x^2} dx = \frac{1}{2} \sqrt{\frac{\pi}{\alpha}}$
$\Delta U = q + w$	$w = -nRT \ln \left(\frac{V_2}{V_1} \right)$	$\int_0^\infty x e^{-\alpha x^2} dx = \frac{1}{2\alpha}$
$H = U + PV$	$w = -P_{\text{ext}} \Delta V$	$\int_0^\infty x^2 e^{-\alpha x^2} dx = \frac{1}{4} \sqrt{\frac{\pi}{\alpha^3}}$
$q = mc_s \Delta T$	$\Delta S_{\text{sys}} = \int_1^2 \frac{dq_{\text{rev}}}{T}$	$N! \approx \left(\frac{N}{e} \right)^N$
$q_{\text{calorimeter}} = C_{\text{calorimeter}} \Delta T$	$\Delta S_{\text{sys}} = \frac{q_{\text{rev}}}{T}$	
$c_P = c_v + R$	$\Delta S_{\text{sys}} = nR \ln \left(\frac{V_2}{V_1} \right)$	
$c_P = \frac{5}{2}R$	$\Delta S_{\text{surr}} = \frac{q_{\text{surr}}}{T_{\text{surr}}}$	$f(x) = \sum_{i=0}^{\infty} \frac{1}{i!} \left. \frac{d^i f(x)}{dx^i} \right _{x=a} (x-a)^i$