

Name and section: _____

1. The partition function, q , can generally be written as

$$q = \sum_i e^{-\beta\epsilon_i} \quad (1)$$

where ϵ_i is the energy of the state i and $\beta = 1/(k_B T)$.

- (a) Derive the average energy in terms of the partition function. Starting from the expression

$$\langle \epsilon \rangle = \sum_i \epsilon_i p(\epsilon_i) = \sum_i \frac{\epsilon_i e^{-\beta\epsilon_i}}{q} \quad (2)$$

where $p(\epsilon_i)$ is the probability of occupying state i . Derive

$$\langle \epsilon \rangle = -\frac{1}{q} \frac{\partial q}{\partial \beta} = -\frac{\partial \ln q}{\partial \beta} \quad (3)$$

using eqs. (1) and (2).

Solution: First, using the chain rule:

$$\frac{\partial \ln q}{\partial \beta} = \frac{1}{q} \frac{\partial q}{\partial \beta}$$

which shows part.

Next,

$$\begin{aligned} -\frac{1}{q} \frac{\partial q}{\partial \beta} &= -\frac{1}{q} \frac{\partial}{\partial \beta} \sum_i e^{-\beta\epsilon_i} \\ &= -\frac{1}{q} \sum_i -\epsilon_i e^{-\beta\epsilon_i} \\ &= \sum_i \frac{\epsilon_i e^{-\beta\epsilon_i}}{q} \end{aligned}$$

- (b) Now let's study the temperature dependence of the average vibrational energy of a diatomic molecule. We learned from CH351 that the vibrational motion of a diatomic molecule can be described as a one-dimensional harmonic oscillator problem. If the molecule vibrates with frequency ω , what are the quantized vibrational energy levels?

Solution:

$$\epsilon_i = \hbar\omega \left(i + \frac{1}{2} \right)$$

- (c) Let's derive the vibrational partition function for this diatomic molecule. Starting with the definition of the partition function given in eq. (1), show that

$$q = \frac{e^{-\beta\hbar\omega/2}}{1 - e^{-\beta\hbar\omega}}. \quad (4)$$

Solution:

$$\begin{aligned} q &= \sum_{i=0}^{\infty} e^{-\beta\hbar\omega(i+\frac{1}{2})} \\ &= e^{-\beta\hbar\omega/2} \sum_{i=0}^{\infty} e^{-\beta\hbar\omega i} \\ &= e^{-\beta\hbar\omega/2} \frac{1}{1 - e^{-\beta\hbar\omega}} \\ q &= \frac{e^{-\beta\hbar\omega/2}}{1 - e^{-\beta\hbar\omega}} \end{aligned}$$

- (d) Show that the average vibrational energy for the diatomic molecule is

$$\langle \epsilon \rangle = \frac{1}{2} \hbar\omega \left(\frac{1 + e^{-\beta\hbar\omega}}{1 - e^{-\beta\hbar\omega}} \right). \quad (5)$$

Solution:

$$\begin{aligned} \langle \epsilon \rangle &= -\frac{\partial}{\partial \beta} \left(-\beta\hbar\omega/2 - \ln(1 - e^{-\beta\hbar\omega}) \right) \\ &= \frac{\hbar\omega}{2} + \frac{\hbar\omega e^{-\beta\hbar\omega}}{1 - e^{-\beta\hbar\omega}} \\ &= \frac{1}{2} \hbar\omega \left(\frac{1 - e^{-\beta\hbar\omega} + 2e^{-\beta\hbar\omega}}{1 - e^{-\beta\hbar\omega}} \right) \\ \langle \epsilon \rangle &= \frac{1}{2} \hbar\omega \left(\frac{1 + e^{-\beta\hbar\omega}}{1 - e^{-\beta\hbar\omega}} \right) \end{aligned}$$

- (e) In the limit as $T \rightarrow 0$, to what value does $\langle \epsilon \rangle$ converge?

Solution: As $T \rightarrow 0$, $\beta \rightarrow \infty$:

$$\lim_{\beta \rightarrow \infty} \langle \epsilon \rangle = \frac{1}{2} \hbar \omega \left(\frac{1}{1} \right)$$

as is expected because of the zero point energy for quantum harmonic oscillators.

- (f) What about the limit of $T \rightarrow \infty$? Taylor series expand the exponential terms to first order in $\beta \hbar \omega$ to estimate the *classical* limit of $\langle \epsilon \rangle$.

Solution: As $T \rightarrow \infty$, $\beta \rightarrow 0$:

Taking the Maclaurin Series of $e^{-\beta \hbar \omega}$,

$$\begin{aligned} e^{-\beta \hbar \omega} &\approx 1 - \beta \hbar \omega \\ \langle \epsilon \rangle &\approx \frac{1}{2} \hbar \omega \left(\frac{2 + \beta \hbar \omega}{\beta \hbar \omega} \right) \\ &= \frac{1}{\beta} + \frac{1}{2} \hbar \omega \\ \lim_{T \rightarrow \infty} \langle \epsilon \rangle &\approx k_B T \end{aligned}$$

- (g) Graphically show how $\langle \epsilon \rangle$ varies with T .

Solution:

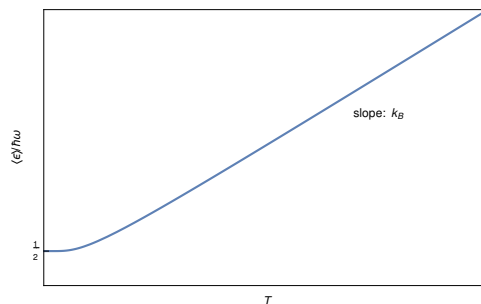


Figure 1: $\langle \epsilon \rangle$ vs. temperature for a quantum harmonic oscillator: At the low temperature limit, the expected energy approaches the zero point energy of the oscillator, but as the temperature rises, the expected energy grows linearly with slope k_B .