

EXP. NUMBER <u>∞</u>	EXPERIMENT/SUBJECT <u>Appropriate notebook tutorial</u>	DATE <u>3/14/15</u>	01
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Objectives

- ☐ Develop good notebook skills
- ☐ practice preparing a notebook
- ☐ ... something else.

Procedure

Observations

1. Break down large steps into small, doable steps.
2. Combine small steps; eliminate redundant info; streamline
3. Measure mass of solid urea delivered to volumetric flask (~ 0.1 g) (50 mL)

urea
white, crystalline
solid.

Mass flask + solid	
Mass flask after	
Mass solid delivered	

4. Boil solution on hotplate $\leftarrow$ Turns green!

~~***~~ Careful: hotplate is hot; glassware looks same hot & cold ~~***~~

5. Add a small amount of water to pipette and drain it. Add a small amount of soap to the pipette and drain it. Rinse the pipette. Rinse well. Rinse with the solution to be transferred, but don't contaminate the solution. Add 10 mL to pipette and drain into a 250-mL Beaker
5. Transfer (vol. pipette) 10 mL of soln to a 250 mL beaker.

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